



Organocatalytic α -amination–allylation-RCM strategy: enantioselective synthesis of cyclic hydrazines

Aram Lim, Jung Hoon Choi, Jinsung Tae*

Department of Chemistry and Center for Bioactive Molecular Hybrids (CBMH), Yonsei University, Seoul 120-749, Republic of Korea

ARTICLE INFO

Article history:

Received 28 April 2008

Revised 30 May 2008

Accepted 2 June 2008

Available online 5 June 2008

Keywords:

Organocatalysts

Cyclic hydrazines

Asymmetric amination

Allylation

Ring-closing metathesis

ABSTRACT

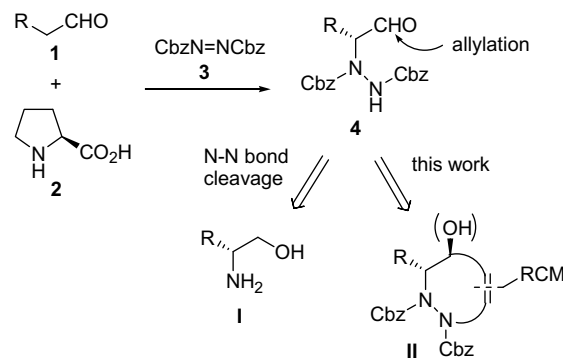
A highly enantioselective method for the synthesis of cyclic hydrazines by using organocatalytic α -amination–allylation-RCM strategy is described. Proline-catalyzed α -amination of aldehydes followed by indium-mediated one-pot allylation of the crude α -hydrazino aldehydes produces 1,2-aminoalcohols in high enantio- and diastereoselectivities. The 1,2-aminoalcohols are further converted into cyclic hydrazines by using ring-closing metathesis (RCM) reaction.

© 2008 Elsevier Ltd. All rights reserved.

1. Introduction

For the past few years, organocatalytic asymmetric synthesis¹ has gained great attention, complementing bio- and metal-catalysis. Several synthetic methods have been developed as practical tools for asymmetric organic synthesis. Especially, the enantioselective α -amination of aldehydes and ketones using proline² and its derivatives has been extensively utilized to introduce amino groups enantioselectively. Cleavage of the N–N bond introduced by dialkyl azidodicarboxylate (RCO₂N=NCO₂R) leads to synthetically useful α -amino alcohol (**I**)² or α -amino acid^{2a} derivatives (Scheme 1). In other ways, the α -amino aldehydes were further utilized in the subsequent tandem reactions such as aldol,³ Passerini,⁴ and Horner–Wadsworth–Emmons reactions.⁵ While allylation reactions of proline-catalyzed aldol products⁶ and aminoxylation adducts⁷ are known, allylation reaction of the organocatalytic α -amination adducts has not been reported up to now. Although few examples have shown that the N–N bond of **4** could be utilized for the synthesis of cyclic hydrazine derivatives; however, they are limited only to 6-membered piperazine derivatives, where the cyclization step involves either intramolecular alkylation reaction⁸ or Wittig reaction.⁹

Herein, we report indium-promoted one-pot allylation of **4** and enantioselective synthesis of cyclic hydrazines (**II**) via an organocatalytic α -amination–allylation-RCM strategy. Synthetic methods for small to medium heterocycles containing an N–O¹⁰ or N–N¹¹



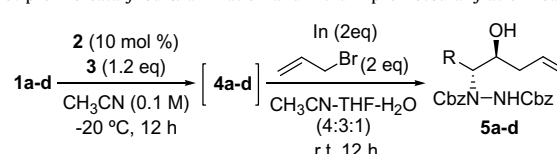
Scheme 1. Strategy for asymmetric synthesis of cyclic hydrazines.

bond by using ring-closing metathesis (RCM) reaction as the key step have been reported in our group. As an asymmetric version of these methodologies, we envisioned that organocatalytic α -hydrazidation reaction followed by RCM could be synthetically useful for enantioselective synthesis of functionalized cyclic hydrazines.

First, we examined the feasibility of the one-pot α -amination–allylation sequence. After treating aldehydes with 10 mol% of L-proline and dibenzyl diazodicarboxylate (CbzN=NCbz, **3**) in acetonitrile at -20°C for 12 h, the reaction mixture was diluted with THF/H₂O (3:1) and then reacted with indium powder and allyl bromide at room temperature for 12 h (Table 1). The overall transformation turned out to be highly enantioselective and

* Corresponding author. Tel.: +82 2 2123 2603; fax: +82 2 364 7050.
E-mail address: jstae@yonsei.ac.kr (J. Tae).

Table 1
One-pot proline-catalyzed α -amination and indium-promoted allylation reactions



Entry	1	5		
		% Yield	dr (<i>anti</i> : <i>syn</i>)	% ee ^a (<i>anti</i> / <i>syn</i>)
1	1a (R = Me)	5a (70)	>99:1	99/(–) ^b
2	1b (R = <i>i</i> -Pr)	5b (88)	98:2	98/(–) ^b
3	1c (R = <i>n</i> -Bu)	5c (67)	81:19	99/>98 ^c
4	1d (R = (CH ₂) ₂ CH=CH ₂)	5d (83)	87:13	98/>98 ^c

^a The % ee's were determined using Chiralcel OD-H column (1% *i*-PrOH in hexanes).

^b The % ee's for the *syn* isomers of **5a** and **5b** were not determined.

^c The minor enantiomers of the *syn* isomers of **5c** and **5d** were not observed.

diastereoselective. When R is small (R = Me, entry 1, Table 1), complete diastereoselective allylation is observed, but relatively bulky R groups show decreased diastereoselectivities (entries 3 and 4, Table 1).

The stereochemistries of the allylation step are determined after conversion of **5** to 1,3-oxazolidin-2-ones where the ¹H NMR chemical shifts of the methine protons¹² were compared with the known compounds.¹³ According to the assigned stereochemistries of **5**, the indium-promoted allylation step proceeds through the Felkin–Ahn transition state (Fig. 1). It seems that the sterically bulky –NCbzNHCBz group prevents chelation of the allylindium reagent to give the Cram product.¹⁴

Next, the RCM substrates were prepared according to the known procedures (Scheme 2). The dienes **8a–c** were prepared using standard alkylation or acylation conditions^{10,11} from **6**, which was synthesized from aldehyde **4d** in 63% yield. Substrates **7a–d** were prepared from **5a–d** in two steps.

The substrates **7a–d** and **8a–c** were reacted with 10 mol % Grubbs' catalysts (**11** or **12**) under refluxing dichloromethane or toluene solution (0.02 M) (Table 2). Sterically bulky substrate (entry 2) and less reactive acryl amide substrate (entry 6) were reacted using 2nd-generation catalyst (**12**). Cyclization yields are good for most substrates to give 8-membered cyclic hydrazine compounds. The triene substrate **7d** yielded 7-membered product **13d** only instead of the corresponding 8-membered cyclic hydrazine (entry 4). Facile ring-closure to cycloheptene rings was further confirmed from the reaction of substrates **9** and **10**, which were prepared from **5d** (Scheme 3). Under the condition A, **9** produced functionalized cycloheptene **13h** in 85% yield in less than 2 h (entry 8). Interestingly, the dieneyne **10** gave only the diene-RCM product **13i** with no enyne-RCM products (entry 9).

In conclusion, we have shown that the indium-promoted in situ allylation of organocatalytic α -amination adducts could introduce 1,2-aminoalcohols with high enantio- and diastereoselectivities. In addition, we have demonstrated that functionalized cyclic hydrazine derivatives could be readily accessed by using the organocatalytic α -amination–allylation–RCM strategy.

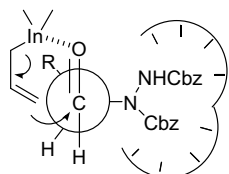
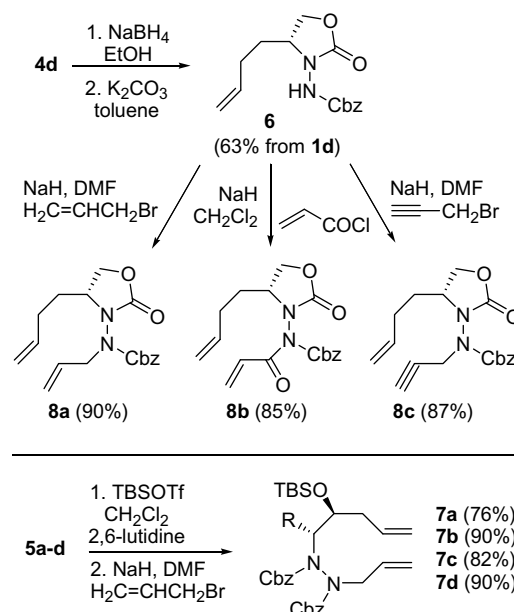


Figure 1. A proposed transition state for the allylation reaction.



Scheme 2. Synthesis of RCM substrates from L-proline-catalyzed α -amination adducts.

2. Experimental procedures

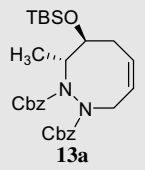
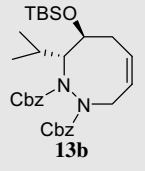
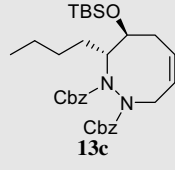
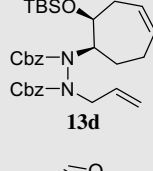
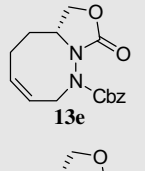
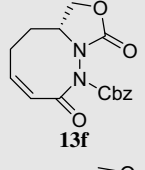
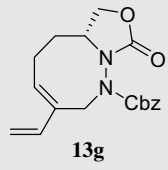
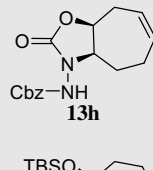
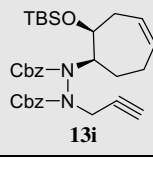
2.1. General procedure for the organocatalytic α -amination–allylation reaction

To a mixture of **1a** (100 mg, 1.72 mmol) and L-proline (18 mg, 0.17 mmol) in acetonitrile (17 mL) was added the dibenzyl azodicarboxylate (616 mg, 2.07 mmol) and the reaction mixture was stirred for 12 h at $-20\text{ }^{\circ}\text{C}$. The reaction mixture was diluted with 17 mL of THF/H₂O (3:1) and indium powder (-100 mesh indium powder purchased from Aldrich; 395 mg, 3.44 mmol) and allyl bromide (2.99 mL, 3.44 mmol) were added. The reaction mixture was stirred for 12 h at room temperature, quenched with saturated 2 N HCl (10 mL), extracted with EtOAc (10 mL \times 3). The combined organic solutions were dried over anhydrous MgSO₄, and the residue was chromatographed on silica gel (hexanes/EtOAc = 2:1) to give **5a** (479 mg, 70%) as a white solid; $R_f = 0.4$ (hexanes/EtOAc = 2:1); mp $74\text{--}76\text{ }^{\circ}\text{C}$; $[\alpha]_D^{20} +2.28$ (c 1.00, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.32–7.25 (m, 10H), 6.71 (br s, 1H), 5.80 (br s, 1H), 5.14–5.08 (m, 7H), 4.19–3.99 (m, 2H), 2.20 (m, 2H), 1.15–1.14 (d, $J = 6.8$ Hz, 3H); ¹³C NMR (62.9 MHz, CDCl₃) δ 156.6, 155.8, 135.8, 135.6, 135.4, 134.8, 128.6, 128.5, 128.4, 127.9, 117.5, 72.4, 68.8, 68.1, 67.9, 57.8, 57.0, 38.1, 29.7, 9.9, 0.0; IR (film, cm⁻¹) 3292, 3062, 3030, 2935, 2355, 1712, 1528, 1498, 1454, 1408, 1262, 1056, 1027; HRMS calcd for C₂₂H₂₆N₂O₅ (M⁺): 398.1842; found: 398.1840.

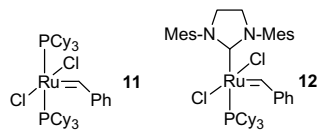
2.2. General procedure for the synthesis of RCM substrates 7

To a solution of **5a** (238 mg, 0.598 mmol) in CH₂Cl₂ (15 mL) were added 2,6-lutidine (0.10 mL, 0.89 mmol) and TBSOTf (0.20 mL, 0.89 mmol) at $0\text{ }^{\circ}\text{C}$ under argon atmosphere. The resulting mixture was stirred at $0\text{ }^{\circ}\text{C}$ for 20 min, warmed to room temperature, and quenched with saturated aqueous NaHCO₃. The aqueous solution was extracted with CH₂Cl₂ (10 mL \times 3) and the combined organic solutions were dried over anhydrous MgSO₄ and the residue was flash chromatographed on silica gel (hexanes/EtOAc = 10:1) to give TBS-protected compound. This product was dissolved in DMF (15 mL) and treated with NaH (47 mg,

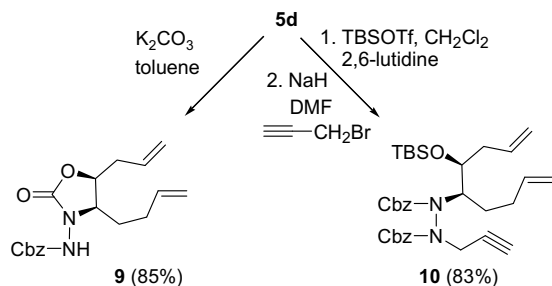
Table 2
RCM reactions of compounds 7–10

Entry	Substrate	Conditions ^a	13	Yield ^b (%)
1	7a	A 4 h	 13a	92
2	7b	12 (10 mol %) toluene reflux, 6 h	 13b	88
3	7c	A 8 h	 13c	59
4	7d	A 2 h	 13d	66
5	8a	A 4 h	 13e	85
6	8b	12 (10 mol %) CH ₂ Cl ₂ reflux, 8 h	 13f	82
7	8c	A 4 h	 13g	75
8	9	A 2 h	 13h	85
9	10	A 2 h	 13i	23

^a Condition **A**: **11** (10 mol %), CH₂Cl₂, reflux, argon.



^b Isolated yields.



Scheme 3. Synthesis of RCM substrates **9** and **10**.

1.2 mmol) followed by allyl bromide (0.10 mL, 1.2 mmol) at 0 °C. The solution was stirred for 30 min at 0 °C and warmed to room temperature for 2.5 h. The reaction mixture was quenched with saturated aqueous NH₄Cl (20 mL), extracted with CH₂Cl₂ (10 mL × 3). The combined organic solutions were dried over anhydrous MgSO₄ and the residue was chromatographed on silica gel (hexanes/EtOAc = 6:1) to give **7a** (250 mg, 76%) as a colorless oil; *R*_f = 0.4 (silica gel, hexanes/EtOAc = 6:1); [α]_D²⁰ +7.33 (c 1.00, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.29–7.23 (m, 10H), 5.89 (m, 1H), 5.26–4.91 (m, 9H), 4.19–3.85 (m, 2H), 2.45–2.03 (m, 2H), 1.25–0.86 (m, 14H), 0.05–0.00 (m, 5H); ¹³C NMR (62.9 MHz, CDCl₃) δ 156.9, 156.4, 135.9, 135.8, 134.4, 133.5, 128.5, 128.2, 128.0, 127.9, 117.4, 72.4, 68.2, 67.9, 67.7, 59.8, 58.7, 56.5, 56.3, 56.0, 39.2, 38.7, 26.0, 18.1, 15.5, 11.6, –3.8, –4.1, –4.3; IR (film, cm^{–1}) 2954, 2919, 2848, 1714, 1454, 1399, 1301, 1223, 1145, 1086, 1043, 914; HRMS calcd for C₃₁H₄₄N₂O₅Si (M⁺): 552.3020; found: 552.3021.

2.3. General procedure for the RCM reaction

A mixture of **7a** (135 mg, 0.240 mmol) and **11** (16 mg, 0.024 mmol) in CH₂Cl₂ (12 mL) was refluxed for 4 h under argon. After concentration of the solution in vacuo, the residue was chromatographed on silica gel (hexanes/EtOAc = 6:1) to give **13a** (115 mg, 92%) as a colorless oil; *R*_f = 0.3 (hexanes/EtOAc = 6:1); [α]_D²⁰ +8.71 (c 1.00, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.45–7.32 (m, 10H), 6.05 (m, 2H), 5.30–5.12 (m, 4H), 4.53–4.32 (m, 2H), 3.70–3.80 (m, 1H), 3.22–3.20 (m, 1H), 2.33 (m, 2H), 1.75–1.15 (m, 3H), 0.96–0.91 (m, 9H), 0.18–0.06 (m, 6H); ¹³C NMR (62.9 MHz, CDCl₃) δ 155.1, 154.1, 136.2, 136.0, 128.7, 128.6, 128.5, 128.2, 128.1, 128.0, 127.9, 127.7, 127.5, 126.4, 68.1, 67.2, 65.7, 43.8, 33.2, 25.8, 18.0, 15.0, –4.3, –4.4, –4.7, –4.8; IR (film, cm^{–1}) 2954, 2923, 2856, 1711, 1462, 1399, 1356, 1305, 1286, 1258, 1231, 1106, 1086, 1027, 933; HRMS calcd for C₂₉H₄₀N₂O₅Si (M⁺): 547.2707; found: 547.2707.

Acknowledgments

This work was supported by the Center for Bioactive Molecular Hybrids (MOST/KOSEF). A.L. thanks BK 21 program (KRF). J.T.

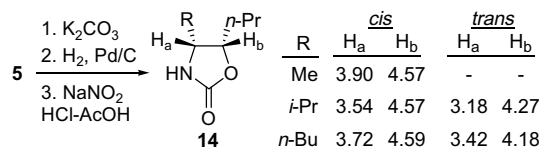
thanks Yonsei University and Professor K.W.J. at USC for the supports during the sabbatical year.

Supplementary data

The ¹H NMR copies of 1,3-oxazolidin-2-ones **14** (see Ref. 12) showing diastereomeric ratios are available. Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2008.06.004.

References and notes

- (a) Gaunt, M. J.; Johansson, C. C. C.; McNally, A.; Vo, N. T. *Drug Discovery Today* **2007**, *12*, 8–27; (b) Akiyama, T.; Itoh, J.; Fuchibe, K. *Adv. Synth. Catal.* **2006**, *348*, 999–1010; (c) Seayad, J.; List, B. *Org. Biomol. Chem.* **2005**, *3*, 719–724; (d) List, B. *Chem. Commun.* **2005**, 719–722; (e) Dalko, P. I.; Moisan, L. *Angew. Chem., Int. Ed.* **2004**, *43*, 5138–5175; (f) List, B. *Acc. Chem. Res.* **2004**, *37*, 548–557; (g) List, B. *Tetrahedron* **2002**, *58*, 5573–5590; (h) Dalko, P. I.; Moisan, L. *Angew. Chem., Int. Ed.* **2001**, *41*, 3726–3748.
- (a) Bøgevig, A.; Juhl, K.; Kumaragurubaran, N.; Zhuang, W.; Jørgensen, K. A. *Angew. Chem., Int. Ed.* **2002**, *41*, 1790–1793; (b) List, B. *J. Am. Chem. Soc.* **2002**, *124*, 5656–5657.
- Chowdari, N. S.; Ramachary, D. B.; Barbas, C. F., III *Org. Lett.* **2003**, *5*, 1685–1688.
- Umbreen, S.; Brockhaus, M.; Ehrenberg, H.; Schmidt, B. *Eur. J. Org. Chem.* **2006**, 4585–4595.
- Kotkar, S. P.; Chavan, V. B.; Sudalai, A. *Org. Lett.* **2007**, *9*, 1001–1004.
- Källström, S.; Erkkilä, A.; Pihko, P. M.; Sjöholm, R.; Sillanpää, R.; Leino, R. *Synlett* **2005**, 751–756.
- Zhong, G. *Chem. Commun.* **2004**, 606–607.
- (a) Henmi, Y.; Kakino, K.; Yoshitomi, Y.; Hara, O.; Hamada, Y. *Tetrahedron: Asymmetry* **2004**, *15*, 3477–3481; (b) Chandrasekhar, S.; Parimala, G.; Tiwari, B.; Narsihmulu, C.; Sarma, G. D. *Synthesis* **2007**, 1677–1682.
- Oelke, A. J.; Kumarn, S.; Longbottom, D. A.; Ley, S. V. *Synlett* **2006**, 2548–2552.
- (a) Yang, Y.-K.; Tae, J. *Synlett* **2003**, 1043–1045; (b) Yang, Y.-K.; Tae, J. *Synlett* **2003**, 2017–2019; (c) Yang, Y.-K.; Lee, S.; Tae, J. *Bull. Korean Chem. Soc.* **2004**, *25*, 1307–1308; (d) Yang, Y.-K.; Choi, J.-H.; Tae, J. *J. Org. Chem.* **2005**, *70*, 6995–6998.
- (a) Tae, J.; Hahn, D.-W. *Tetrahedron Lett.* **2004**, *45*, 3757–3760; (b) Kim, Y. J.; Lee, D. *Org. Lett.* **2004**, *6*, 4351–4354.
- The diastereomeric ratios were determined after conversion of **5** to 1,3-oxazolidin-2-ones **14** (for the procedures of the cleavage of the N–N bond in step 3, see: Baumann, T.; Vogt, H.; Bräse, S. *Eur. J. Org. Chem.* **2007**, 266–282). While the chemical shifts of H_a and H_b for *cis*-isomers in **14** appear at 3.5–3.9 ppm and 4.5–4.7 ppm, respectively, those of *trans*-isomers appear at 3.2–3.4 ppm and 4.2–4.3 ppm (see Ref. 13 and Supplementary data).



- (a) Kano, S.; Yokomatsu, T.; Iwasawa, H.; Shibuya, S. *Chem. Lett.* **1987**, 1531–1534; (b) Hu, N. X.; Aso, Y.; Otsubo, T.; Ogura, F. *J. Org. Chem.* **1989**, *54*, 4398–4404; (c) Bach, T.; Schlummer, B.; Harms, K. *Chem. Eur. J.* **2001**, *7*, 2581–2594.
- Paquette, L. A.; Mitzel, T. M.; Isaac, M. B.; Crasto, C. F.; Schomer, W. W. *J. Org. Chem.* **1997**, *62*, 4293–4301.